Table 3. *Vibration parameters*

The temperature factor for the heavy atoms was: $\exp[-2\pi^2(h^2 a^{*2} U_{11}+2h k a^* b^* U_{12}+\cdots)]$.
The temperature factor for the hydrogen atoms was: $\exp[-(B\sin^2\theta/\lambda^2)]$.

	U_{11}	U_{22}	U_{33}	U_{12}	U_{13}	U_{23}
N(1)	0·0396 (6)	0·0312 (8)	0·0578 (10)	−0·0069 (6)	−0·0136 (5)	−0·0043 (9)
N(2)	0·0480 (8)	0·0281 (8)	0·0635 (10)	0·0025 (6)	−0·0127 (7)	−0·0017 (9)
O(1)	0·0331 (5)	0·0257 (8)	0·0537 (7)	−0·0022 (3)	−0·0129 (3)	−0·0030 (4)
O(2)	0·0377 (5)	0·0234 (8)	0·0602 (7)	−0·0016 (3)	−0·0093 (5)	0·0013 (4)
O(3)	0·0289 (5)	0·0288 (8)	0·0498 (7)	−0·0006 (3)	−0·0096 (3)	0·0035 (4)
C(1)	0·0317 (6)	0·0281 (8)	0·0347 (7)	−0·0022 (6)	0·0007 (5)	−0·0022 (4)
C(2)	0·0266 (5)	0·0257 (16)	0·0344 (7)	0·0025 (6)	0·0009 (5)	−0·0004 (4)
H(1)	7·3 (6)					
H(2)	4·8 (4)					
H(3)	4·9 (5)					
H(4)	6·1 (6)					
H(5)	4·9 (5)					

Table 4. *Bond distances* (Å)

	Urea–oxalic acid (*a*), (*d*), (*g*)	Oxalic acid dihydrate (*b*), (*e*), (*g*)	Urea (*c*), (*e*), (*f*)
C(2)–C(2′)	1·521 (5)	1·536 (3)	
C(2)–O(2)	1·209 (3)	1·212 (4)	
C(2)–O(3)	1·289 (5)	1·291 (5)	
O(3)–H(1)	0·96 (4)	1·026 (7)	
C(1)–N(1)	1·318 (6)		1·352 (2)
C(1)–N(2)	1·321 (4)		
C(1)–O(1)	1·260 (3)		1.260 (3)
N(2)–H(3)	0·86 (2)		1·003 (4)
N(1)–H(4)	0·84 (2)		
N(2)–H(2)	0·86 (4)		0·998 (5)
N(1)–H(5)	0·88 (3)		
O(1)···O(3)	2·472 (5)		
O(2)···N(2)	2·999 (5)		
O(3″)···N(2)	3·123 (3)		
O(2‴)···N(1)	2·911 (3)		
O(1′)···N(1)	2·941 (4)		

(*a*) Present work.
(*b*) Sabine, Cox & Craven (1969).
(*c*) Pryor & Sanger (1970).
(*d*) X-ray.
(*e*) Neutron.
(*f*) Bond lengths corrected for thermal motion.
(*g*) Bond lengths not corrected for thermal motion.

Table 5. *Bond angles* (°)

	Urea-oxalic acid (*a*), (*d*), (*g*)	Oxalic acid dihydrate (*b*), (*e*), (*g*)	Urea (*c*), (*e*), (*f*)
C(2′)–C(2)–O(2)	121·8 (2)	121·0 (3)	121·7 (1)
C(2′)–C(2)–O(3)	112·3 (3)	112·4 (3)	
O(2)–C(2)–O(3)	125·9 (2)	126·6 (3)	
C(2)–O(3)–H(1)	110 (3)	114·4 (6)	
O(1)–C(1)–N(1)	119·8 (3)		121·7 (1)
O(1)–C(1)–N(2)	122·2 (2)		
N(1)–C(1)–N(2)	118·1 (3)		116·6 (1)
C(1)–N(1)–H(5)	118 (3)		119·0 (3)
C(1)–N(2)–H(2)	118 (2)		
C(1)–N(1)–H(4)	122 (3)		120·2 (3)
C(1)–N(2)–H(3)	123 (2)		
H(4)–N(1)–H(5)	120 (3)		
H(2)–N(2)–H(3)	118 (4)		

(*a*) Present work.
(*b*) Sabine, Cox & Craven (1969).
(*c*) Pryor & Sanger (1970).
(*d*) X-ray.
(*e*) Neutron.
(*f*) Bond lengths corrected for thermal motion.
(*g*) Bond lengths not corrected for thermal motion.

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Structural studies of metal dithiocarbamates. V. The crystal and molecular structure of bis-(*N*,*N*-diethyldithiocarbamato)tellurium(II)

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(*Received* 9 *November* 1971)

A correction to a reference in *Acta Cryst.* (1971) B**27**, 1499 is given.

In a paper of the above title (Fabiani, Spagna, Vaciago & Zambonelli, 1971) the reference to the structure of lead(II) bis-(*O*,*O*-diisopropylphosphorodithioate) is incorrect. On page 1504, column 1, line 20 (Guzy, Raynor & Stodulski, 1969) should read (Lawton & Kokotailo, 1969) and in the list of references GUZY, C. M., RAYNOR, J. B. & STODULSKI, L. P. (1969). *Nature, Lond.* **221**, 550 should read LAWTON, S. L. & KOKOTAILO, G. T. (1969). *Nature, Lond.* **221**, 550.

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